

# **Sample Preparation**

# for the Laser Particle Sizer ANALYSETTE 22



Sample preparation

- 1. Sample division
- 2. Sample preparation
  - 2.1 Dispersion
  - 2.2 Tips and tricks
- 3. Materials and suitable measuring liquids



#### Sample preparation

It is always astounding that the market for highly precise, fully automatic analytical instruments is permanently growing, while no emphasis is placed on the equally important sample preparation or sampling.

One of the most common mistakes during analysis is already made during the sampling stage and drawn into the analytical procedure right from the beginning. Again and again, it surprises process technicians how often samples are drawn carelessly and how the achieved analysis results are readily accepted.

Initially the capabilities of the measuring instrument are often doubted when repeat measurements deviate, but the source of the error is deeper though: in an inhomogeneous sampling.

Therefor the results are only reproducible if the analysed sample *represents* the goods to be tested with a high degree of exactness, i.e. the sample taken can be equated with the entire batch.

### 1. Sample division

The sample for the Laser Particle Sizer ANALYSETTE 22 (approximately 200 mg - 1 g) shall correspond in the contained particle type and particle distribution with the entire batch of material.

The **Rotary Cone Sample Divider LABORETTE 27** is very well suited for the division of dry laboratory samples or suspensions, since different dividing heads with different division ratios can be selected. Depending on the division head up to 3000 dividing steps per minute can be achieved.

Possible division: 1:8, 1:10 and 1:30

Example: a laboratory sample of 50 ml (g) respectively 100 ml (g), could initially be divided with the dividing head 1:10 and then with the dividing head 1:30, so that for the Laser Particle Sizer ANALYSETTE 22 required amount of 200-400 mg is available.

Now the **representative** sample will be deglomerated.

### 2. Sample preparation

Via pre-tests it must determined in which manner the sample material can be moistened and dispersed.

The liquid should possibly completely and spontaneously wet the solid matter.

The additional ultrasonic support (if possible with maximum power) in general reduces the dispersion duration.



- The particles of the solid matter shall be individually and free of agglomerates in the suspension.
- The dispersion condition must be *stable* during the entire measurement, coagulating / flocculating of the particles may not occur.
- Coagulating in a suspension can be recognized by a slight, slushy bottom deposit, which will swim in the sample glass at slight movements of the glass, as a second phase respectively layer and displays a *cloudlike* shape.
- Floating of the sample material on the surface of the liquid is a sign of non-wetted sample material, i.e. the already added share of dispersants like tensides/ wetting agents or salts is too low.
- The sample may not be destroyed during the dispersion, respectively comminuted. With thin, platelet shaped material like mica, kaolin, clay and inorganic salts this is especially important.

Here it is also recommended to paste a small sample amount on a specimen holder with a little bit of liquid and wetting agent. Now under the microscope the particle spectrum and the maximum particle size can be determined.

After the finished dispersion it is also microscopically checked if the coarse particles are still present or were destroyed.

- The vehicle liquid/ measuring liquid must in all cases have a smaller, max. equal specific weight (density) than the solid material to be measured.
- The sample may not partially dissolve, dissolve or swell!

### 2.1 Dispersion

Unproblematic samples, which submerge without any great effort directly into the surface of the water and possess no large amount of fine share, are added to the dispersion unit as a *solid material* with a spatula portion by portion and after a brief ultrasonic treatment/ dispersion (30-60s) measured reproducibly.

Shows the conducted *double measurement* too large differences in the particle size distribution it may be attributed to several reasons:

- pump speed too low the coarse material deposits
- the agitator speed is too great air bubbles are created, respectively air is stirred in
- dispersion duration is too brief



- Increasing fine share
  - Longer ultrasonic treatment necessary
  - Addition of dispersion aids
  - Fine particles stick to the measuring cell glass
  - Clogging due to coarse particle >2 mm
  - Coarse material deposits: specific weight too high
- Seemingly coarser appearing curve
  - Sample swells or flocculates
  - Sample agglomerates
  - Sample is magnetic
- Declining coarse- and fine range
  - Sample dissolves
  - Beam absorption drops

Additionally faulty measurements due to jammed or bent hose connections can occur.

Samples hard to disperse may display the following characteristics: static charge (for example plastics), adhesive forces or cohesive powers – samples tend to conglutinate/ agglutinate – (for example clays, soil samples, kaoline), magnetism, hydrophobic characteristics – water repellent molecule components (for example drugs, medications, toner, graphite, titanium dioxide, waxes), coagulation (for example clays, kaoline, chalk, gypsum).

a) Static charge respectively also hydrophobic characteristics:

Here a spatula amount of the material should be added to a small 50 ml Erlenmeyer flask and then at <u>first</u> 1 (up to 2) drops of a wetting agent (surfactant or diluted surfactant solution) added, then mixed into a paste until the sample is completely wetted. Water is added drop by drop and stirred. The now created suspension (approximately 20 - 30 ml) is dispersed in an ultrasonic bath.

If the sample is already inside the dispersion unit and floats on top of the surface, the sample can be wetted as follows:

With a glass rod or with the tip of a spatula, a *small* drop of the wetting agent (for example Dusazin 901, Teepol, Tween 80 or dish soap) is added/ touched on the surface of the liquid and distributed. Immediately it can be seen, that the formed skin on the surface breaks open and the fine portion enters the suspension.

### *b) Adhesive powers:*

are *clinging* powers of the particle. A reduction of these surface powers can be obtained by creating on the boundary phase areas from solid/ liquid, adsorption layers from surface-active agents or macromolecules. It is considered a covering, a guarding or masking of the solid material respectively wetting. For this reason in most cases for example tetra-Na-diphosphat (sodium pyrophosphate: Na4P2O7) or poly sodium approximately at 0.5-1% are utilized.



### c) Cohesive powers and magnetism:

The effects of the forces of attraction between atoms or molecules of a body are described as cohesive power or polarity respectively magnetic characteristics. The magnetic characteristics are difficult to eliminate: with only slight magnetism a highly viscous liquid like for example ethylene glycol or a glycerine/ water mixture may be used or the sample is heated above several 100°C. This is hardly possible in a laboratory and seldom realisable. Therefor are such materials with a high degree of magnetism are not suitable for the particle size analysis.

### d) Coagulation:

is the flocculation of a sample caused by the agglomeration of colloid particles in a suspension. This can happen with too great amounts of solids or an unfavourable ph-range. By adding several drops of undiluted acid (for example hydrochloric acid) prior to adding the sample!! for the acidly range or through diluted lies (for example caustic lye, ammonia or even soda solution) in the alkaline range, may the ph-value be lowered respectively increased, so it counteracts the reaction of the sample (for example with chalk, kaolin, hydrated lime and clay). Suitable are here also Na2HPO4 (alkaline) or KH2PO4 (acidy) as a 0.1-1% solution.

### 2.2 Tips and Tricks

The larger the fine share of a sample, the greater is the dispersion effort. A necessary ultrasonic treatment lasting several minutes (or even longer) should be conducted in an external ultrasonic bath.

Here it is recommended to add to a 50 ml Erlenmeyer flask a spatula tip full (approximately 0.5-1 g) of sample and via simple wetting adding approximately 20 ml of the measuring liquid + the dissolved/ mixed dispersion aid (surfactant).

We recommend when utilizing surfactants those with low foam tendencies like for example Dusazin 901.

After briefly shaking the Erlenmeyer flask, it is fastened with laboratory clip inside the ultrasonic bath so the inner level of liquid is below the surface of the liquid of the ultrasonic bath.

After during the pre-tests determined deagglomeration period, the necessary suspension amount for the measurement will be added with a shaking motion via a pipette into the dispersion unit of the Laser Particle Sizer ANALYSETTE 22.

Is the needed amount of solids exactly known, it may be weighed directly into the Erlenmeyer flask and after the dispersion the *complete* contents can be conveyed with a wash bottle, so no separation due to sampling with a pipette occurs.

With a few small tricks difficult samples like for example fly ash, sulphur, coal, plastics or pigments can be quickly dispersed even in water.



The sample is mixed with **one** drop of surfactant and after adding **one to two** drops of water with a glass stirrer or spatula mixed to a *paste*. Due to the relatively high share of surfactant compared to the share of water the cavitation is reduced and the sample quickly wetted.

After further addition of several drops of water and stirring at the same time it can be checked if not already wetted particles are swimming on top of the surface of the liquid.

Now the suspension is diluted down to approximately 20-30 ml and deagglomerated in the ultrasonic bath.

When using simple surfactants like dish soap it often occurs that when *pasting* and mixing the sample "foam" evolves which after the dispersion swims on top of the surface and is traversed into the particle sizer.

A too high stirring intensity pulls the foam into the measuring circuit and *"coarse particles"* are measured which are not present.

In order to avoid this "faulty measurement" it is possible to destroy the *"foam"* inside the Erlenmeyer flask: the tip of a glass stirrer is dipped into <u>*n*-butanol</u> so the glass stirrer is just wetted but no drops are recognizable.

By slightly touching the surface of the foam it collapses and the suspension can easily be analysed for particle size.

An additional possibility – with poor wetting – would be the addition of 2 to 3 drops of alcohol (for example ethyl alcohol) directly on the dry laboratory sample, which immediately absorbs the alcohol like a sponge. Now again water and a wetting agent can be added and dispersed accordingly.

The user, working with the "Small Volume Wet Dispersion Unit" of course has the possibility to utilize alcohol, alkanes, high-boiling benzines or other organic liquids.

It should be mentioned, that the Small Volume Wet Dispersion Unit is **not** explosion protected – please select the corresponding liquids – and only operate in well ventilated areas.

The exchange of the measurement liquid for example alcohols to other organic liquids is relatively easy, since many liquids can be mixed easily amongst each other.

The compatibility of solvents with the connective hoses should also be kept in mind. The seals in the measuring cell and the connective hoses are made of Viton. Acetone (ketones), acetates and enamel thinner cannot be used.

The *resistance lists* available from hose manufacturers contain for the most common elastomers a rating of the chemical resistance against various operating mediums (liquids).

After a conducted measurement with the ANALYSETTE 22 it should become habit to directly rinse



the measuring system in order to avoid an unnecessary "depositing" or "adhering" of particles in the measuring circuit and especially on the measuring cell.

An "in-between rinsing" with a surfactant is very helpful.

Residue on the measurement cell may be caused by tap water respectively limy water. Either it should be switched to distilled water or the measurement cell has to be cleaned from time to time.

The <u>lime residue</u> is removed in a few minutes by rinsing with 10% hydrochlorid acid. Afterwards it should be rinsed twice with regular water.

On the following pages you find a list of materials with the suitable dispersion liquids and additives. Please note the list is only available in German. Should you have questions please contact the FRITSCH laboratory: gerber@fritsch.de



# Materials and suitable measuring liquids

Peststoff		Sedimentations- flüssigkeit	Zusatz	setoff
Årt	Dichte g/om <sup>3</sup>	11400184910	Art	Konzentration g/l
Askerboden (s.s.Boden, Erde)		Wasser Wasser Wasser	Tetranatri unpyrophosphat Natriumoxalat Ammoniakwasser	0,451,35 0,67 5,8 Vol. %
Artivkohle (w.a.Kohle)	2,0	Wasser Isopropanol Wasser Wasser Wasser	Ammoniakwasser - Tetranstriumpyrophosphat Natriumlinoleat Natriumoxalat	5,8 Yol. % - o.A. o.A. o.A.
Alaunerde	1,8	Wasser Wasser Wasser Wasser Wasser Tetrachlorkohlenstoff	- Natriumtartrat Natriumozalat Natriumhezametaphosphat Sulzsäure -	- 1,0 ohne Angabe 1,0 ohne Angabe
Alaunerdezement	1	Ethylenglycol	Cobaltchlorid	ohne Angabe
Alkalisalze		Leinöl + Xylol n-Butylamin Cyclohexanon Cylohexanol n-Butanol		
Aluminium •	2,7	Tetrachlorkohlenstoff Wasser + 50 Vol.% Bthylenglycol Wasser	-	-
		Wesser Wasser Wasser	Natriumhexametaphosphat In triumtarprat Natriumoxalat	8.A. 0.A.
		Wasser + Ethylenglycol Bthylenglycol Cyclohexanon Cyclohexanol Chloroform Isopropanol Wasser	Trinatriumphosphat Trinatriumphosphat - - Salzsäure	o.A. o.A. - - pH = 3
Aluminiumfluorid		Ethylenglycol Ethylenglycol Ethylenglycol	Calciumchlorid Strontiumchlorid Cobaltchlorid	0,05 = 0,5 0,05 = 0,5 0,05 = 0,5
luminiumhydroxid	2,32,4		- Sacoharose	40 \$
Aluminiumozid (s.s. Tonerde, Korund	3,54,1	n-Butylamin Tetrachlorkohlenstoff Wasser Wasser Wasser Wasser Wasser	- Natriumherametaphosphat Natriumoarbonat Tetranatriumpyrophosphat Kalium/Natriumherameta- phosphat Natriumtartrat	- 0,5 - 1,0 0.A. 0,31,5 1,0 1,0
	Carlo March	Wasser n-Butanol Cycloheranon	Salzsäure	pH = 3



Peststoff		Sedimentations-	Zusatzstoff	
Art	Dichte g/cm <sup>3</sup>	flüssigkeit	,Art ,	Konsentration g/l
Aluminiumsilioid		Wasser Wasser	Tetranatriumpyrophosphat Trinatriumphosphat	0,31,5 0.A.
Annoniumperchlorat	2,0	Isobutanol Benzol	- Naphthalinstearosulfosäur	einige Tropfer
Anhydrit		Methanol	-	-
Anthracenpaste	1,2	Wasser	Trinatriumphosphat	o.A. '
Anthracit	1,41,7	Wasser Wasser	Trinatriumphosphat Natriumalkylnaphthalen- sulfonat	0,5 1,0
Antimonozid .	3,85,3	Wasser Wasser Wasser	Tetranatriumpyrophosphat Kalium/Natriumhexameta- phosphat Natriumhexametaphosphat	0,3 - 1,5 0,5 0.A.
Apatit		Wasser Wasser	Trins triumphosphat Aluminiumchlorid	7,8
Arsenate (nicht wasser- löslich)	-	Wasser	Tetranatriumpyrophosphat	0,451,35
Arsenige Säure		Cyclohexanon n-Octanol	-	:
Arsentrioxid	3,8	n-Octanol Cyclohexanol Petroleum	- - Ölsäure	-
Asche (s.a.Flugasche, Ki Werksasche)	raft-	Wasser	Tetranstriumpyrophosphat	1,0
Bariumoarbonat	4,4	Cyclohexanen Methanol Wasser	- - Tetranatriumpyrophosphat	- - 1,0
Bariumsalze (nicht wass löslich)	er-	Wasser .	Tetranatriumpyrophosphat	0,451,35
Bariumstrontiumoarbonst		Wasser + Ethanol Wasser +Methanol Cyclohexanon	<u>.</u>	-
Bariumsulfat, Baryt	4,3 4,5	Wasser Wasser Wasser Wasser Wasser + Ethylenglycol Wasser Wasser + Methanol	Alkylphenolethylenorid- Kondensat Tetranatriumpyrophosphat Tetranatriumpyrophosphat + Selzsäure Trinatriumphosphat - Natriumherametaphosphat -	1,0 0,32,25 0,31,5 3,65 0.A. - 0,5
Bariumtitanat	5,35,8	Wasser Wasser Cyclohexanon Wasser + Ethylenglycol	Natriumhexametaphosphat Kalium/Natriumhexameta- phosphat - -	0,5 1,0 -
Baurit	3,3	Wasser	Tetranatriumpyrophosphat	c.A.

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Feststoff		Sedimentations- flüssigkeit	Zusatz	stoff
Art	Dichte g/cm <sup>3</sup>		Art K	onzentration g/l
Bentonit	2,7	Wasser : Wasser Wasser Wasser Wasser Wasser	Natriumoarbonat Ammoniak Natronlauge Tetranatriumpyrophosphat Natriumozalat Kalium/Natriumhexameta- phosphat Natriumsilikat (Wassergl	0,05 1,0
Berlinerblau		Wasser	Tetranatriumpyrophosphat	0,31,5
Beryll		Wasser Wasser	Natriumsilikat Natriumherametaphosphat	0.A. 0.A.
Binsetein	5.1	Wasser	-	-
Bismutverbindungen	·	Wasser	Tetranatriumpyrophosphat	1,35
Blanc fire		Wasser	Tetranatriumpyrophosphat	0,9
Blei	11,3	Wasser Aceton Cyclohexan Cyclohexanol Cyclohexanon Isoamylalkohol		-
Bleicherie		Wasser Wasser Wasser	Trinatriumphosphat Tetranatriumpyrophosphat -	o.A. 0,451,35 -
Bleicyanamid		Wasser	Tetranatriumpyrophosphat	0,31,5
Bleifarben (s.a.Bleioxid, Mennige)		Wasser Cyclohexanol	Tetranatriumpyrophosphat -	0,451,35
Bleioxide (s.a.Bleifarben, Mennige)	89,5	Ethylenglycol Wasser Xylol Cycloheranon Paraffinöl + Benzol Wasser	- Tetranatriumpyrophosphat - - Natriumhexametaphosphat Kalium/Natriumhexameta- phosphat	- 0,51,5 - - 0,5 1
Bleisulfat	5,6	Wasser	Trinatriumphosphat	0.4.
Bleisulfid	7,3	Cyolohexanol	-	-
Boden (s.s.Aokerboden, Erde)		Wasser Wasser Butylphthalat + Ethanol	- Batriumozalat	- 0,6720 g/l -
Boroarbid	2,5	Tauser	Tetranatriumpyrophosphat	o.A.
Braunkohle (s.a.Kohle)	<sup>1,2.</sup> 1,4	Wasser Isobutanol Diethylphthalat Cycloheranon + 10 Masse-% Methanol Cycloheranol + 10 Masse-% Methanol	Netzmittel - - -	einige Tropfen - -



Peststoff		Sedimentations- flüssigkeit	Zusstzstoff	
Art	Dichte g/cm <sup>3</sup>	TIMOTERAL	Art	Konzentration g/l
Breunstein	4,9	Wasser Wasser + Ethylenglycol	Tetranatriumpyrophosphet Trinatriumphosphat	0,451,35
Bronze	8,78,9	Cyclohexanol Cyclohexanon	-	-
Cadmiumarsenat	4,2	Wasser + 50 Vol% Methanol	-	-
Cadmiumfarben		Wasser Wasser	Natriumhexametaphosphat Tetranatriumpyrophosphat	1,0 0,451,35
Cadmiumsulfid		Wasser Ethylenglycol	Tetranatriumpyrophosphat -	o.A. -
Calciumarsenat .		Wasser + 50 Vol% Ethanol	-	-
		Wasser + 50 Vol% Nethanol	-	-
Calciumcarbonat, Kalk- spat (s.s.Kreide)	2,7-2,9	Ethylenglycol Wasser	-	-
		Wasser Wasser Wasser	Ammoniakwasser Trinstriumphosphat Tetranatriumpyrophosphat	5,8 Vol% o.A. 0,32,25
е Т		Wasser Wasser Wasser	Natriumsilikat (Wasserglas) Natriumbexametaphosphat	
8 N N R		Wasser	Natriumsilikat (Wasserglas) + Kaliumcitrat	
131 <sub>13</sub>		Wasser	Nstriumsilikat + Tetranatriumpyrophosphat	1:0
. * s <sup>3</sup>		Wasser Wasser	Natriumcitrat + Tetranatriumpyrophosphat Natriummethylendinaphthyl-	1,0 1,0 1,0
		Xylol	sulfonat	-
		Wasser + 20 Vol% Glycerin Cyclohexanon +	-	
· · ·		10 Vol% Isoamyl- alkohol	-	
2		Wasser + 50 Vol% Ethylenglycol	• • • • • • •	-
aloiumfluorid, Flußspat	3,2	Wasser Wasser Wasser	Ammoniakwasser Keliumchlorid Selpetersäure	1 Vol% 0,074 0,126 (0,002 n)
		Wasser Wasser	Tetranatriumpyrophosphat Gelatine + Natriumcarbonat	0,31,5
		Methanol Cyclohexanol	+ Fatriumoarbonat Kaliumchlorid	0,074
9 2		Cyclohersnon Aceton		-
aloiumhydroxid	2,3	Cyclohexanol	-	-
		Ethanol Isopropanol	<b>.</b>	-
		Wasser	Natriumhexametaphosphat	0,5 .



Peststoff	a manufilm	Sedimentations- flüssigkeit	Zusatzsto:	ff -
Art	Diohte g/om <sup>3</sup>		Art .	Konzentration g/l
Calcium-Magnesium- carbonat, Dolomit	2,9	Wasser Wasser Wasser	Trinatriumphosphat Tetranatriumpyrophosphat Ammoniak	o. A. 0,31,5 5,8 Vol. %
Calciumphosphat	2,3	Bthylenglycol Chinolin Aceton Cyclohexanon Cyclohexanol Sthylenglycol Petroleum Ethylenglycol Ethylenglycol Ethylenglycol Isobutanol	- - - - Calciumchlorid Strontiumchlorid Cobaltohlorid	- - - - 0,050,5 0,050,5 0,050,5
(wasserlöslich)		Hexan n-Octanol	1	<b>1</b>
Calciumphosphat (nicht wasserlösl.)	2,23,2	Wasser Wasser Wasser Wasser Wasser Wasser + Ethanol n-Butanol Ethanol	- Tetranatriumpyrophosphat Natriumhexametaphosphat Natriumsilikat (Wasser- glas) Trinatriumphosphat	- o.A. 0,51,0 1,0 o.A. - -
Calciumsalse (nicht wa	sserlöslich)	Wasser	Tetranatriumpyrophosphat	0,9
Calciumstannat		Wasser	Tetranatriumpyrophosphat	1,0
Calciumsulfat (s.u.Gip	s, Anhydrit)			
Calciumwolframat		Wasser	Natriumcitrat	0,5
Carborundum (s. Silici	umcarbid)			0
Cellulose		Benzin Testbenzin Benzol	- Trinatriumphosphat	- 1,0
Cerussit		Wasser .	Natriumhexametaphosphat	0.A.
hina clay	1.1.1	Tasser	Trinatriumphosphat	-
Chrom	646.24	Isobutanol	-	-
Chromfarben		Wasser	Tetranatriumpyrophosphat	0,451,35
Chrongelb	3,3	Cyclohexanon		-
Chromozi d	2,75,3	Wasser Cyclohexanol + 10 Vol Isosaylalkohol Wasser	Tetranatriumpyrophosphat .%- Kalium/Natriumhexameta- phosphat * Natriumcarbonat	0,32,25 - 0,6 0,12
Cobalt (s. Kobalt)				
Cordierit	3,0	Wasser	Natriumsalz der polymeri- sierten Carboxylsäure	einige Tropfen einer wäßrigen Lösung



Peststoff	GC	Sedimentations-	Zuset	zstoff
Art	Dichte g/cm <sup>3</sup>	flüssigkeit	Art	Konzentration g/l
Diamant	3,5	Olivenöl	-	-
× .	1.0	Tasser	Gelatine	1,02,0
		Wasser	+ Natriumcarbonat Trinstriumphosphst	pH = 9
1 M (c) 1 M	1 S. R. I	Rthanol	-	0.4.
		Wasser	Natriumhexametaphosphat	0.5
latomesnerde		Wasser	Natriumhexametaphosphat	
		Wasser	-	-
Mealoiumphosphat	2,3	Wethanol	_	-
	-15	Wasser + Ethanol	<ul> <li></li></ul>	
olomit (s. Calcium/ agnesiumcarbonat)	2,9			
lisen	7,8	Cyclohexan	-	-
		Wasser + Ethylen-Glycol	-	-
이 문화 '한 아이'	1.4	Sojabl + 50 Vol. %	in a le j	-
	4 4 Q <sup>20</sup> 1	Aceton		
· · · · ·		Rüböl + Aceton		
18		Wasser + Ethylenglycol Cyclohexanol	Trinatriumphosphat	0.A.
1 and 1	Sec. 5. 1	Cyclohexanon	-	
1		Wasner	Tetranatriumpyro-	1.0
11 C K		and the second second	phosphat	
isenoxide	3,45,7	Wasser	Kalium/Natriumhexameta- phosphat	0,5
10 H (10 H)		Wasser	Tetranatriumpyrophos- phat	0,31,5
677 - 6	19. L	Paraffin51 + Benzol	- 1	-
- 177 - H		Wasser	Natriumhexametaphosphat	0,5
isenoxidrot	1.4.4.4	Wasser	Tetranatriumpyro- phosphat	1,0
1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1		Iylenollösung	•	-
isenschwarz	nicht mögl	ich, da magnetische Floc	tung	
isensulfat	1,83,0	Isobutanol	-	-
isensulfid (s.s. Pyrit)	4,8	Cyclohexanon		•
mail		Wasser	Tetranstriumpyro- phosphat	0,31,5
nstatit	3,03,3	Wasser	Kalium (Natriumhexameta	-1,0
	t of	Veccov	phosphat Natriumcarbonat	o.A.
	1.1.1	Wasser Wasser	Tetranatriumpyro-	1,0
	- 14 A	HABOUL .	phosphat	110
rde		Wasser	-	-
s.s.Boden, Ackerboden)	1994 S. 199	Wasser	Tetranstriumpyro-	0,45 - 1,35
	The eff		phosphat	
		Wasser Wasser + Ethylenglycol	Natriumoxalat	20,0
10 N	а <sup>т</sup> , <sup>т</sup>	Butylphthalat + Etha- nol		
arben, mineralische		Wasser	Kaliumoitrat	30
	8	Wasser	Tetranatriumpyro-	0,31,5
eldspat	2,6	Wasser	-	-
oraspar	2,0	Wasser	Trinatriumphosphat	o.A.
		Wasser	Natriumoxelat	o.A.



Peatatoff		Şedimentations- flüssigkeit	Zusatzstoff	
Art	Dichte g/om <sup>3</sup>	TIMBERRET	Art	Konzentration g/1
PeSiCr	4,9	Cyclohexanol Cyclohexanon	-	:
Flint	2,7	Wasser Wasser Wasser	- Natriumoxalat Tetranatriumpyro- phosphat	 o.A. 0,451,35
Flugasche (s.a.Asche, Kraftworksasche)	2,2-2,3	Wasser Wasser Wasser	- Tetranatriumpyro- phosphat Na-Salz der polymeri- sierten substituierten Alkylbenzolsulfon- säure	- 0,451,35 0,57
Fluoride		Wasser	Natriumcarbonst + Gelatine	12,5
FluSepat (s.Calcium- fluorid)				3
Formsand		Vasser Vasser Vasser	Natronlauge Tetranatriumpyrophos- phat Trinatriumphosphat	o.A. 0,451,35 o.A.
Porsterit	-	Wasser	Tetranatriumpyro- phosphat	1,0
Fritten		Wasser Wasser	- Tetranatriumpyro- phospht Natronlauge	- 0,451,35 0.4.
Füller		Wasser	Trinstriumphosphat	0.A.
Getreidemehl	1,5	Isobutanol Isobutanol + Diethyl- phthalat Diethylphthalat Petroleum	-	:
Glpe	2,3	Ethanol Sthylenglycol Ethanol Methanol n-Amylalkohol Methanol + Ethylengly- col + Ethanol	Calciumchlorid Cobaltcitrat - - Galciumchlorid	10 o.A. - - o.A.
		Ethylenglycol Ethylenglycol Ethylenglycol Ethylenglycol	Cobaltoitrat Calciumchlorid Strontiumchlorid Cobaltohlorid	o.A. 0,050,5 0,050,5 0,050,5
Gips (Stuck-)	3,0	Ethylenglycol + 50 Vol.5 Ethanol Ethylenglycol + 50 Vol.5 Ethanol Ethylenglycol	Natriumoitrat Calciumoitrat Cobaltcitrat	1,29 0,5 o.A.
· · · · · · · · · · · · · · · · · · ·	1.000	Me thanol	-	
	2,3			



Feststoff Sedimentations- flüssigkeit			Zusatzstoff	
Årt	Dichte g/cm <sup>3</sup>	fluesigkeit	Art	Konzentration g/1
Glas	2,43,0	Wasser	Tetranatriumpyro- phosphat	0,451,35
		Wasser	Trinstriumphosphat	0.A.
and the former of the	1 . 4	Wasser	-	-
		Wasser	Kalium/Natriumbexameta- phosphat + Natriumcarbonat	0,3 0,06
		Butanol	-	-
		Cyclohexanol	-	-
	1	Wasser + Ethylenglycol	- u u u u u	-
	1 .	Methanol	-	-
	1	Ethylenglycol	-	1-
		Ligninlösung	-	-
A COMPANY AND A COMPANY	and the second	Wasser	Natriumhe xame tapho sphat	0,5
lasuren	1.	Wasser	Tetranstriumpyrophos- phat	0,31,5
Glimmer	2,8	Wasser	Tetranatriumpyrophos- phat	o.A.
Granat	3,8-3,9	Wasser	Natriumhexametaphosphat	0,5
Graphit	2,02,5	Wasser	Gerbsliure	0.5
94. 1	1	Wasser	Natriumlinoleat	5.0
96 a A		Wasser .	Ammoniak + Hatriumlinoleat	0,83,2
	1	Wasser	Trinstriumphosphat	o.A.
	1	Wasser	Ligninsmitonat	10
		Ethanol Wasser Wasser	Carboryme My/ccin/esc Dioctylester der Natrium sulfobernsteinsäure	70.0 0,5 Vol.%
Nmatit	5,2	Wasser	-	-
an a		Wesser	Tetranatriumpyrophos- phat	1,0
exachloroyclohexan	1	Wasser	Trinstriumphosphat	0.A.
lochofenschlacke	2,53,0	Wasser	Natriumhexame taphosphat	1,0
0 12 <b>1</b> 12 12 12		Wasser	Tetrenstriumpyrophos-	0,451,35
0.040		Chinolin	phat	
		Cyclohexanol	🖕 e 👔 ostal 👘 👘	-
		Cyclohexanon		• ·
		Isopropanol	-	-
lydrargillit	2,4	Wasser	Tetranatriumpyrophosphat	1,0
ilmenit	4.7	Wasser	-	-
lagrun	-	Wasser + Ethylenglycol	-	-
akao	1,5	Diethylphthalat	•	-
		Isobutanol	- J.	-
	1	Benzol	- 1	<del>.</del>
	4 4	Isobutancl +		
		Diethylphthalat	5 a a a	
	1. 1	Aceton		- s
		Cyclohexanon	-	-
aliumchlorat .	2,3	Cyclohexanon	-	-



Peststo	ff	Sedimentations-	Zusatz	atoff
Art	Dichte g/cm <sup>3</sup>	flüssigkeit	Art	Konzentration g/l
Kalkhydrat (s.Caloiumhydroxid)				
Kalkstein (s. Calcium- carbonat, Kreide)				
Kalomel .	7,2	Cyclohexanon Cyclohexanol		2
Kaolin	2,22,6	Wasser	Ammoniak	0,2
•	SS	Wasser	Trinatriumphosphat	o.A.
		Wasser	Salzaäure	pH'= 3
		Wasser	Tetranatriumpyro- phosphat	0,32,25
2	1	Wasser	Tetranatriumpyro-	2,25
			+ Natriunsilikat (Winner-	1,0
		Wasser	Kalium/Natriumhexameta-	1,0
	10	Wasser	Kalium/Natriumhexameta- phosphat	
	2 s		+ Natriumcarbonat	0,127
		Wasser	Natriummetaphosphat + Natriumcarbonat	0.45 0,011
		Wasser	Natriumozalat	0,67
		Wasser	Natronlauge	einige Tropfen
		Wasser	Natriumcarbonat	0,5
1		Wasser	Natriumsilikat (Wasserglas)	0,2 - 1
· · · ·	1	Wasser	Natriampolyácrylat	10
[artoffelmehl		Isobutanol	-	-
		Cyclohexanon	-	-
	-	Isobutanol +		-
		Diethylphthalat		
		Diethylphthalat		-
feramische Massen		Wasser	Tetranatriumpyro- phosphat	0,451,4
		Wasser	Kalium/Natriumhexa- metaphosphat	1.0
81.0		Wasser	Kalium/Natriumhexameta- phosphat	0.k.
4			+ Batriumcarbonat	o.A.
lieselgur		Wasser	-	-
	•	Wasser	Natriumsilikat (Wasserglas)	0.4.
a - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -		Wasser	Natriumozalat	0.67
		Wasser	Trinatriumphosphat	0.A.
		Wesser	Amnoniak	2,0
				and the second
lieselgut		Wasser		
	2,22,3	Wasser	Tetranatriumpyro- phosphat	1.0
Kleie		Wasser	Trinatriumphosphat	o.A.
Knoohenasche		Weaser	-	2



	1.4.4	
-	10	-
	nv	-

Pestato	ii	Sedimentations- flüssigkeit	Zusatzstoff	
Art	Dichte g/cm <sup>3</sup>		Art	Konzentration g/l
Kobalt	8,8	Isobutanol	-	-
		Cyclohexan	-0. 11 8	-
		Cyclohexanon	-	-
		Diethylphthalat	-	-
		Cyclohexanol	NG 11 11 12 13 13	-
		Ethanol		
		Rubbl + Aceton	-	- 3
12		Wasser + Ethylenglyco	lTrinstriumphosphat	0.A.
- X		Ethanol + 5 % Wasser	-	-
Kobaltorid		Wasser	Kalium/Natriumhexameta-	0,6
			phosphat + Natriumcarbonat	0,12
Kohle (s.s.Aktivkohle, Braunkohle, Steinkohle)		Wasser	Calciumchlorid	1,0
Braunkohle, Steinkohle)		Ethanol	Calciumchlorid	1015
**		Ethylenglycol	Calciumchlorid	0,050,5
	1.	Ethylenglycol	Strontiumchlorid	0.05 0.5
	1.10	Ethylenglycol	Cobaltchlorid	0.05 0.5
		Cyclohexan	-	-
		Cyclohexanol	<b>2</b> 0 9	-
	i n ne	Cyclohexanon	-	-
		Sthanol	-	-
		Petroleum	ölsäure	1 10
		Benzin	Ölskure	110
		Cyclohexanol + 50 Vol.% Methanol	-	-
Koka	1,6 1,9	Isobutanol	-	-
		Wasser	Natriumalkylnaphthalen- sulfonat	1,0
		Wasser	Natriumlincleat	1.0
		Wasser	Natriumoleat	10.0
	1 S I	Wasser	Gerbaäure	0.5
			+ Anmoniak	0,83.2
	8 C.	Ethanol	Calciumchlorid	1.0
		Ethanol + 50 Vol.%	Calciumchlorid	1,0
		Ethylenglycol		
Korund (s.a.Aluminium- oxid, Tonerde)	4,0	Wasser	-	-
		Wasser Wasser	Tetranstriumpyrophosphat	0,451,35
V			Trinatriumphosphat	o.A.
Kraftwerksasche (s.a. Neche, Flugasche)		Wasser	Tetranatriumpyrophosphat	1,0
Kreide (s.a. Calcium-	2,6	Wasser	-	-
carbonat)	1.000	Wasser	Natriumsilikat (Wasserglas)	2,0
	1 - I	Wasser	Kaliumoitrat	5.5
		Wasser	Tetranstriumpyrophosphat	0,31,5
		Wasser	Ammoniak	2,0
		Aceton	-	- 10 M
	Sec. Tage	Petroleum		-
		Isopropanol	-	-
Kreide (gefällt)		Isopropanol	-	-
Gryolith	3,0	Wasser + 20 Vol.% Glycerin	-	-
		Ethylenglycol		
		Sthylenglycol Wasser	Potenatalumpurchashet	
		wasser Ethylenglycol	Tetranstriumpyrophosphat	-
Kunststoffe		Wasser	Tetranatriumpyrophosphat	0,451,35
A MALE VE SUITE	1			
		Wasser	Trinatriumphosphat	0.A.



Peststoff		Sedimentations- flüssigkeit	Zussat	astoff
Art	Dichte g/cm <sup>3</sup>		Art	Konsentration g/l
Kupfer	8,9	Janser	-	-
		Tosor	Tetranatriaapyrophosphat	20
		Aceton	-	-
		R5551		-
	6 e	RUb31 + Ageton	-	-
		Sojačl + 50 Vol.S Aceton	-	-
		Cyclohexanon	-	-
	2	Cyclohemmol		-
	Constant and the	fsoamylalkohol		-
Kupferhydroxid		Jasser	Hatriumiexanotaphosphat	0,5
Kupferoxychlorid		Wneser	-	-
Kupferphthalocyanin		Janser		-
Kupferschlacke		Jasser	fetranstriuspyrophosphet	1,0
Kupferverbindungen		Jasoer	Tetrunatriumpyrophosphat	0,451,35
(nicht wasserlösl.)		Jaaser	Entriwaherasetsphosphat	0,5
Läppulver		Janser	Trinstriumphosphat	0.4.
Leuchtstoffe	-	Janner	Entriumcitrat	0,5
Lignit		Inobutanol	-	-
		Dicthylphthulat	2 N	
		Cyclohernsol + 10 % Lethanol	-	-
Lithopene	4.2	Diethylphthulat	-	-
		Glyceein		-
. ¥		Jan: er	Estriusmethylendinaphthyl- culfornt	1,0
		Jassor	Tetranstriumpyrophosphot	0,31,5
20 A 1		Acsaer	Frinktriumphosphat	0.1.
		Wnsser + 33 % Glycerin	-	
688		Jusper	Trinatriumphonghat	o.A.
1		Vances	Annoulak'	0,1
		Wagger	Natriumallikat (Wasserglas)	0.1.



Peststoff		Sedimentations-	Zusa	tastoff
Art	Dichte g/cm <sup>3</sup>	flüssigkeit	Årt	Konzentration g/1
Magnesiumcarbonat	3,5	Wasser	Tetranatriumpyrophosphat	0,31,5
		Methanol	-	-
1.1		Wasser	Ammoniskwasser	5.8 Vol. %
		Cyclohexanon	-	-
		Sthylenglycol	Calciumchlorid .	0,050,5
		Ethylenglycol	Strontiumchlorid	0,050,5
	1	Ethylenglycol Ethylenglycol	Cobaltchlorid	0,050,5
		renkienRikcor		
Magnesiumoxid	2,836	Ethylenglycol	-	-
	1	Methanol		-
Magnesiumsilicid	í	Wasser	Tetranatriumpyrophosphat	0,31,5
Magneeiumsilikate (s.s. Enstatit)	-3,03,3	Wasser ·	Kalium/Natriumhexameta- phosphat	1,0
		Wasser	Natriumcarbonat	e.A.
10 E E	1	Wasser	Tetranatriumpyrophosphat	1,0
		Wasser	Kalium/Natriumherameta-	0.A.
			phosphat + Natriumcarbonat	0.4.
		-		1
Magnetit		Wasser		
	i i	Ethanol Methanol		1 -
	6 63	Nitrobenzol		1 2
Mangan		Cyclohexanon	16 J	
		Isobutanol		
Mangancarbonat		Wasser	Kalium/Natriumhexameta- phosphat	1,0
		Wasser	Tetranatriumpyrophosphat	1.0
- <sup>1</sup> - 1		Cyclohexanon	-	
Mangahdioxid, Pyro- lysit	4,74,8	Wasser	Tetranatriumpyrophosphat	0,32,25
Manganoxide	4,55,4	Wasser	Tetranatriumpyrophosphat	0,32,25
Wehl		Petroleum	Ölsäure	1,010,0
1. 4 - X		Isobutanol	1 x <sup>15</sup>	-
2 (H		Isobutanol		-
N 186 0.874	1 1 N 1 1	+ Diethylphthalat		1
8 19 17	1. 1. 1.	Diethylphthalat Benzin		1 0 0
97.5		Benzol		1 2
		Cycloheranon	-	1 -
50 I.		Benzin	Ölsäure	1,010,0
terreture ( a s. Mart	0.0	Paraffinöl + Benzol		
fennige (s.a. Blei- oxide, Bleifarben)	9,0	Cyclohexanon	- I	1 1 -
		Ethylenglycol		-
-		Wasser + Ethylen-	Trinatriumphosphat	0.4.
	20	glgcol	n na ser na ser se	
		Cyclohexanol	Patament of second second second	
		Wasser	Tetranatriumpyrophosphat	0,31,5
		Xylol Ethylenglycol	Calciumchlorid	0,050,5
	1 an	Ethylenglycol	Strontiumchlorid	0.050.5
				0.050,5

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Feststoff	CC 1981 1995 1995	Sedimentations-	Zusatzstoff	
	1	flüssigkeit		
Árt	Dichte g/om <sup>3</sup>	1	Årt	Konsentration g/1
Mergel	2,7	Wasser	Tetranatriumpyrophosphat	0.4.
		Wasser	Kalium/Natriumhexameta- phosphat	1,0
		Wasser	Kalium/Natriumhexameta-	0.4.
	10 K		phosphat + Natriumcarbonat	0.4.
Metalle (s. direkt um	ter den Bl	ementen)		
Methylmethaorylat		Wasser	-	-
Wilchpulver	1,4	n-Octanol		
i i cupui vvi		Isobutanol	1	
Miloriblau		Wasser	Trinstriumphosphat	o.A.
Mineralfarben	2	Wasser	Kaliumoitrat	30,6
Mineralwolle		Cyclohexanon	-	-
alybdän	10,2	Ethanol		1
no sy buent	1010	Aceton	-	
		Olycerin	1	_
		Wasser + Glycerin	-	-
	c1)	Wasser + Ethylengly-	-	-
•	8	col Ethylenglycol	-	_
		the day Edward		
Wolybdünsulfid		Cycloheranon	-	-
Natriumbicarbonat	2,2	Cyclohexanon	-	-
	-	Cyclohexanol		-
Tatriumphosphat		Ethanol		
fickel	8,8	Cyclohexanon + 10 % Aceton	1	0.7
		Cyclohexan	-	14 IV
		Cyclohexanon	-	-
		Cyclohexanol		-
		Rüböl + Aceton	-	-
		Wasser + Glyzerin	-	-
lickelox14	6,8	Wasser + Glycerin	-	-
Organische Pulver		Isobutanol +	-	-
		Diethylphthalat n-Octanol		
		Isoanylalkohol		
Penicillin	1. 1.1	Isooctan		
	-	1		
Petrol eunkoks		Wethanol Wasser	-	
Phosphate (s.a. Roh- phosphate)			Tetranatriumpyrophosphat Natriumhexametaphosphat	0,9
Phosphor (rot)	2,2	Ethanol		-
		Wasser	Kaliumsilikat	0,12 Vol. \$
		Wasser	Natriumhexametaphosphat	0,5
Phosphor (w418)	1,8	Wasser	Natriumsals der polymeri- sierten, substituierten Akylbesolsulfonsäure + Kaliumsilikat	0.2



Peststoff				
Art	Dichte g/cm <sup>3</sup>	Sedimentations- flüssigkeit	Art	Konzentration g/l
Pigmente		Cyclohexanon		-
		Cyclohexanol	-	
1 (a) (b)		Isopropanol	_	-
	1	Wasper	Tetrenatriumpyrophosphat	0,452,25
		Wasser + Ethylenglycol	-	-
Polymethylmetacrylat		Wasser	Trinstriumphosphat '	-
Polyvinylacetat		Wasser /	Tetranatriumpyrophosphat	o.A.
Folyvinylchlorid	1,4	Isopropanol	Natriumlinolest	0.A.
		Isobutanol	-	-
		Wasner	Gerballure	1.0
1.00	1	Wasser	Trinatriumphosphat	o. A.
	1.00	Washer	Natriumlinolest	o.A.
Polyester		Paraffinöl	-	
Portlandzement	3,1	n-Butanol		-
	- 5554 - 55	Benzylalkohol		-
	1	Chinolin	-	-
		Cyclohexanol	-	-
	1	Cyclohexanon	-	-
1		Cyclohexanon + Iso-		-
	-	anylalkohol		1.00
5 B	1	Ethylenglycol		2 - S
	1	Ethanol	C-lciumchlorid	0,050,2
	1 1 1 E	Ethanol	Strontlumchlorid ·	0,080,3
1		Ethylenglycol	Calciume+lor1d	0,110,45
		Ethylenglycol	Strontiumchlorid	0,110,45
		Ethanol	-	
1	1	Isobutanol	-	-
		Rizinusöl		10 <b>.</b> - 11
		Steinöl		-
	-1	Methanol	Tetranatriumpyrophosphat	genütligt
		Paraffinöl	-	-
orzellanpulver	2,4	Wesser	Natriumhexametaphosphat	0.5
umicit		Wagger		-
uzzolane	1	Wasser	Tetranatriumpyrophosphat	0,451.35
	. 1	Weaser		-
yrit (s.s. Eisen-	4.4	Ethylenglycol	Calciumchlorid	0,050,5
sulfid)		Ethylenglycol	Strontiumchlorid	0.050.5
		Ethylenglycol	Cobaltchlorid	0.050.5
		Methanol + Tetra-		
14	1	chlorkohlenstoff		
		Wasser Wasser	Tetranatriumpyrophosphat	0.4.
		Wasser + Glycerin	Tetranatriumpyrophosphat	o.A.
uarz (s.s. Sand, Sandstein)	2,65	Wasser	Tetranatriumpyrophosphat	0,451,35
sandstein)		Wasser	Natriumhexametaphosphat	0,5
ł		Wasser	Natriumoxalat ·	0,67
1		Wasser ,	Trinatriumphosphat	0.4.
	1	Wesser Wesser	OH-Ionen	pH = 78
· · · · · · · · · · · · · · · · · · ·			wes - A Grives	



Peststoff Art	Dichte g/cm	Sedimentations- flüssigkeit	2usatz Art	stoff Konzentration g/1
Quecksilberver- bindungen (nicht wasserlösl.)	1	Wasser	Tetranatri umpyrophosphat	0,451,35
Resin		Wasser	Trinatriumphosphat	o.A.
Rohmehl, Rohschlämme (für Zement)		Cyclohexanol Cyclohexanol + 50 Vol. % Iscamyl- alkohol Isobutanol		-
		Wasser Wasser	- Tetranatriumpyrophosphat	0.451.35
Rohphosphate (s.a. Phosphate)		Wasser Wasser	Tetranatriumpyrophosphat Trinatriumphosphat	0,451,35 0.A.
Rohrzucker (s.a. Zucker	1,6	Disthylphthalat Isobutanol Isoamylalkohol	-	-
Ruð	1,72,0	Acetoń Nethanol Vasser	- Dioctylester der Natrium-	-
		Wasser Wasser	sulfobernsteinsüure Natriuslinolest Gerbshure	10.0 1,0
Rutheniumoxid	7.0	Wasser	Natriumhexametaphosphat	0,5
Jand (s.a. Quars, Sandstein)		Ethanol + Butyl- phthalat Wasser Wasser Wasser	- Natriumsilikat (Vassergla Trinatriumphosphat	- a) 2,0 o.A.
Sandstein (s.a. Quarz, Sand)	r a Sacial ca	Wasser Wasser Wasser + Cyclohexanon	Tetranatriumpyrophosphat Trinatriumphosphat	0,451.35 0.A. -
Schamotte	2,6	Wasser Wasser Wasser	Tetranatriumpyrophosphat Kalium/Natriumhexameta- phosphat Kalium/Natriumhexameta- phosphat + Natriumcarbona	0,31,5 1,0 0.4. 0.A.
· · · · · · · · · · · · · · · · · · ·	1.00	Wasser	Natriunnexametaphosphat	0.4.
Schiefer	2,7	Ethanol Wagger	Calciumchlorid Tetranatriumpyrophosphat	0.A. 1,0
Schlacke (s.a. Hoch- ofenschlacke)		Wasser Isopropanol		1
Schleifmittel (s.a. Si- iciumcarbid, Korund)		Wasser Wasser Wasser	Tetranatriumpyrophosphat Tetranatriumpyrophosphat Trinatriumphosphat	0,31,5 0,31,5 0.A.
Schwefel	2,1	Wasser	Natriumlinoleat	0.A. 0.A.
Schwefelktes		Bthylenglycol	-	-
Schwermetallver-		Wasser	Tetranatriumpyrophosphat	0,451,35



Pestatoff		Sedimentations-	Zusatzstoff	
Årt	Dichte	flüssigkeit	Årt	Ronzentration g/l
Selen	4.5	Cyclohexanon	- 19.4	· - · ·
		Cyclohexanol	-	1 SO 🖌 24
		Bthylenglycol	1	
		Ethylenglycol	Calciumchlorid	0,050,5
		Ethylenglycol	Strontiumchlorid	0,050,5
		Ethylenglycol	Cobaltchlorid .	0,050,5
		Waccer	Tetranstriumpyrophosphat	0,451,35
Silberhalogenid	6,0	Vasser	Natriumhexametaphosphat	0,5
Silber - Palladium- paste	10,6	Toluol	-	-
Silicium .	2,4	Wasser	-	-
	15365	Wasser	Natriumhexametaphosphat	0,5
Siliciumcarbiđ	3,2	Methanol	Natriumsalz der Ethylen- diamintetrasseigskure	10,0
		Wasser	-	-
4		Wasser	Kalium/Natriumhexameta-	1,0
	1.1	Wasser	Kalium/Natriumhexameta-	0.A.
151 J	10.00	L RUK DA S	+ Natriumcarbonat	0.A.
		Wasser	Tetranatriumpyrophosphat	0,451,35
	1 8	Wasser	Natriumhexametaphosphat	0,5
23		Wasser	Trinstriumphosphat	0.A.
	1 A. A.	Wasser + Ethylen-	Tetranatriumpyrophosphat	0.4.
	a . * *	glycol Wasser	Nonylphenoxypolyethanol	einige Tropfer
Siliciumoxid		Wasser + 50 Vol. \$	-	-
(s.a. Quarz, Kiesel- gut)	- 83 -	Xylol		Í.
such		Wanger	Natriumhexametaphosphat	0,5
		Vasser	9-10 Ethoxy-Octylphenol	einige Tropfer
		Wasser + Ethanol		-
Silikate (nicht wasserlösl.)		Wagper	-	-
(HIGH'S WESSELIDET.)		Wasser	Tetranatriumpyrophosphat	0,452,25
	1. S.	Wasser + 50 Vol. % Ethanol	-	-
		Wasser + 50 Vol. % Sthylenglycol	-	-
		Wasser + Ethylen- glycol	Trinatriumphosphat	o.A.
		Wasser	Tetranstriumpyrophosphat	2.25
	17. 		+ Natriumoxalat + Natriumhexametaphosphat	2.25 0.67 1.0
Sillimanit	1. The s	Vageer	•	-
V. States	1.000	Wasser	Tetrana triumpyrophosphat	24,8
	12.7%	Wasser + 50 Vol. % Ethanol		-
Stahl	7,8	Wasser + Ethylengly- col		-
		Wasser + 50 Massa \$	Cobaltchlorid	0,1
3 33		Ethylenclycol	· · · ·	
874 J. M. S.	S	Vasaer	Alkylphenolethylenoxid- Kondensat	1,0



Pestatoff		Sedimentations-	Zusatzstoff	
Årt	Dichte g/cm <sup>3</sup>	flüseigkeit	Årt	Konzentration
Stärke	1,5	Isobutanol	-	-
		Diethylphthalat Isobutanol	-	
8		+ Diethylphthalat	1 1 N 1 -	-
		Benzol Methanol	•	-
Steatit				
atestit	2,72,8	Wasser	Kalium/Natriumhexameta- phosphat	1,0
		Wasser	Kalium/Natriumhexameta- phosphat + Kaliumcarbonat	0.A.
· ·		Wasser	Tetranatriumpyrophosphat	-
Steinkohle			· · · · · · · · · · · · · · · · · · ·	
(s.a. Kohle)	1,4	Aceton Cyclohexancl		-
		Cyclohexanol	-	
26		+ 50 Vol. % Methanol	_	-
		Cyclohemanon	-	-
		Cyclohexanon + Metha- nol	-	-
15		Ethylenglycol	-	-
		Ethanol .	Calciuschlorid	11.0
		Sthanol	-	-
al el		Methanol	-	
601		Petroleum	-	-
8 0 E	S	Wasser	Dioctylester der Natrium- sulfobernsteinsäure	5,010,0
		Wasser + Ethanol	-	
÷		Wasser + Ethanol	Natriumlinoleat + Calciumchlorid	0.A. 0.A.
		Wasser + 50 Vol. % 1,3 Butylenglycol	-	-
	1.1	Wasser + 50 Vol. % 1.3 Butylenglycol	Natriumoitrat	0.362
· · · · ·	0 82 8	Wasser + 50 Vol. \$	Merrunderne	0.302
		1,3 Butylenglycol	Netzmittel	0,20,3
	5 8 5	Wasser	Natriumlinolest	10,0
a (1)			+ sulfoniertes Lorol (Hauptanteil Dodecyl- alkohol)	0.A.
		Wasser	Gerbsäure.	0.4.
		Xylol	-	-
trontiumcarbonat	3,7	Cyclohexapon	_	-
		Wasser	Tetranatriumpyrophosphat	1,0
trontiumsalze nicht wasserlösl.)		Wasser	Tetranatriumpyrophosphat	0,451,35
Strontiumtitanat		Yasser	Kalium/Natriumhexameta- phosphat	1,0
Sulfide		Ethylenglycol	14 to (=1) (+1)	-
		Wasser	Saponin	0.4.
ulfonanid	1,3	Isopropanol	9-10 Ethoxycotylphenol	einige Tropfer
falkum -	2.7	Wasser Wasser	Tetranatri umpyrophosphat Natri umbexameta phosphat	0,31,5
fantal	16,6	Cyclohexanol		-
		Cyclohexanon	-	-
		Sthylenglycol	-	0,1



Peststoff		Sedimentations-	Zuentzetoff	
Art	Dichte g/om <sup>3</sup>	flüssigkeit	Art	Konzentration g/l
Thorium		Wasser 33 % Glycerin		-
Titan	5,0	Wasser Wasser Wasser + 50 Masse ≸ Ethylenglycol	Tetranatriumpyrophosphat Natriumhexametaphosphat	1,0 1,0 -
Titancarbid	3,8	Wasser Wasser	Tetranatriumpyrophosphat Natriumsalz der polymeri- sierten Carboxylsäure	1,0 einige Tropfen einer 25 %igen wäser. Lösung
Titandioxid (Rutil, Anatas)	3,84,2	Leinöl Wasser Wasser + 50 Masse ≯	Kalium/Natriumhexameta- phosphat + Kaliumhydroxid	- 0,3 pH = 10,7 -
	· .	Ethylenglycol Wasser	Natriumsalz der polymeri- sierten Carboxylsäure	einige Tropfen einer 25 %igen wässr. Lösung
		Wasser Wasser Cyclohexanon Cyclohexanol + 10 Vol. % Isoamylalkohol Ethylenglycol Xylol	Tetranstriumpyrophosphat Natriumhexametaphosphat - - -	1,0 0,51,0 - - -
Titaneisen		Wasser	-	-
Titanweiß		Wasser Xylenollösung	Tetranstriumpyrophosphat	o.A.
Toluidinrot		Tasser	Trinatriumphosphat	o.A.
Ton	2,52,6	Wasser Wasser Wasser	Natriumcarbonat Ratriumozalat	0.67 .
		Wasser Wasser Wasser	Natriumsilikat (Wasserglas) Natriumhexametaphosphat Natriumpyrophosphat Trinatriumphosphat	20,0 1,0 0,31,5 0.A.
		Wasser Buthylphthslat +	Kalium/Natriumhexameta- phosphat -	1.0
Tonerde		Ethanol Wosser Wasser	Natriumpolyacrylat	40
(s.a. Korund, Aluminiumoxid)		Wasser	Kalium/Natriumhexameta- phosphat	1,0
·		Wasser Wasser Wasser Wasser	Natriumhexametaphosphat Natriumtartrat Salzsäure Tetranatriumpyro-	1,0 1,0 pH = 3 0,31,5
		*sager	phosphat	of strategy

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Feststoff Sedimentations- flüssigkeit		Zueatzstoff		
Årt	Dichte g/cm <sup>3</sup>	liussigkeit	Art	Konsentration 8/1
Tonerdesement (s.a. Zement)	3,2	Sthylenglycol Sthylenglycol Cyclohexanol Chinolin	Calciumchlorid Cobaltchlorid	1,0 0,65
Tonschiefer (s.a. Schiefer)		Wasser	Trinstriumphosphat .	0.4.
Traß		Isobutanol	· · · -	-
Tricalciumphosphat		Wasser Methanol Wasser	Tetranatriumpyrophosphat - -	0,451,35
Tripolyphosphat		Methanol	14	
Trockenhefe		Wasser + Ethanol	-	-
Tuff (vulkan.)		Wasser Wasser Wasser	Natriumozalat Ammoniak Natriumsilikat	0,67 2,0 0.A.
Ultramarin	2,3	Wasser Ethylenglycol	Tetranstriumpyrophosphat	o.A.
Uranerz	• 7,3	Tasser	Natriumhexametaphosphat	0,5
Uranox14	7,1 - 11,0	Isobutanol Wasser Wasser Wasser Wasser Wasser + Glycerin	Trinatriumphosphat Tetranatriumpyrophasphat Natriumhexametaphosphat Natriumslz der polymeri- sierten, substituterten Alkylbenzolsulfonsäure	- 0.A. 1,0 0,5 10,0
Waschpulver		Wasser + Ethylen- glycol		-
Weicheisen (s.s. Bisen)	7,8	Wasser + Glycerin		-
Weizenmehl (s.a. Mehl)	1,5	Cyclohexanon Ethanol Diethylphthalat Isobutanol Isobutanol + Diethylphthalat Fetroleum		-
Wismutverbindungen (s.	u. Bismutv	rbindungen)		
Wolfram	19,1	Wasser Glycerin Aceton + Rüböl Ethanol Aceton	Bthoxyliertes Nonylphenol - - - -	einige Tropfen - - - -
		Methanol Wasser (Peststoff vorher in HP be- handeln und waschen) Wasser + Bthylen-		



Peststoff		Sedimentations-	Zusatzstoff	
Årt	Dichte g/cm <sup>3</sup>	flüssigkeit	Art	Konsentration
Wolfram	19,1	Wasser .	Sacharose + ethoxyliertes Nonyl- phenol	300 einige Tropfen
		Ethylenglycol	Calciumchlorid	0,050,5
- 1982 - C	1. 1. A. 1. S.	Ethylenglycol	Strontiumchlorid	0,050,5
		Ethylenglycol	Cobaltchlorid	0,050,5
Colframearbid .	15,9	Pflanzenöl	-	-
		Wasser Wasser + Ethylen-	-	
		clycol	-	-
•	1 0 1	Ethylenglycol	-	
	1	Ethylenglycol	Calciumchlorid	0,05 0,5
	1	Ethylenglycol	Strontiumchlorid	0,050,5
		Ethylenglycol	Cobsltchlorid	0,050,5
Volframoxide	7.2	Wasser	Tetranstriumpyrophosphat	1,0
	14.1	Cyclohexanon	-	-
		Wasser + Glycerin		
Zahnzement		Ethylenglycol	-	-
Sement_	2,93,2	Pyridin		-
(s.s. Fortland-		Petroleum	Ölsäure	1,010,0
(ement)		Benzol	-	
	- 1	Isopropanol	-	-
		Methanol	Tetranatriumpyrophosphat	gesättigt
	요 문제 전	n-Butanol	-	
	12 8	Isobutanol Paraffinöl		8
10 g		Cyclohexanol		
		Cycloherenol '	1- I -	
е. 1. – С		+ 50 Vol. % Iso- anylalkohol		C 6 6
1.5	1 1	Chinolin	historia de la companya de la	100000 000
		Ethanol	Calciumchlorid	0,0555,5
		Ethanol Ethylenglycol	Strontiumchlorid	0,0750,32
4 <sup>11</sup> 11	S	Ethylenglycol	Calciumchlorid	0,050,5
· · · · · · · · · · · · · · · · · · ·		Sthylenglycol	Strutiunchlorid	0.050,5
	100	Ethylenglycol	Cobaltchlorid	0.050.5
	1.1.4	Benzin		
		Benzin	Gleäure	1,010,0
	1.0151	Methanol + Glycerin		•
ink	7,1	Ethanol	_	11
1997 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 - 1988 -	1.001.004	n-Butanol	and a second second	
	(18) A.	Aceton		-
		Cyclohexanon	1	•
		Cyclohexanol	-	•
		Ethylenglycol	Calciumchlorid	o.A.
		Isobutanol		-
inkoxid	5,55,8	Wasser	Natriumhexametaphosphat	0,51,0
		Wasser	Tetranatriumpyrophosphat	0,31,5
		Wasser	Trinatriumphosphat Kalium/Natriumhexameta-	0.A. 1.0
	100	Wagser	phosphat	
입 것이 하는 것이		Wasser	Natriumsalz der konden- sierten Naphthalinsulfon-	5.0

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			21.	
Peststoff Årt	Dichte g/cm <sup>3</sup>	Sedimentations- flüssigkeit	Zus Art	Atzstoff Konzentration g/1
Zinksulfid	24 <sup>1</sup>	Cyclohexanon	•	-
Zinkweiß		Wasser	Tetranatriumpyrophosphat	0,9
Zinn	7,3	n-Butanol Isobutanol Isobutanol + n-Butanol		-
Zinndioxid	7,0	Wasser Wasser Wasser Wasser	Tetranatriumpyrophosphat Natriumcitrat Kalium/Natriumherameta- phosphat Kalium/Natriumhexameta- phosphat + Natriumcarbonat	0,9 o.A. 1,0 0,6 0,12
Zinnweiß		Wasser	Tetranatriumpyrophosphat	0,451,35
2irkon	6,5	Nethanol Wasser Wasser + 50 Vol. ≸ Nethanol Wasser Wasser Isobutanol	Salzsäure Tetranatriumpyrophosphat - Kalium/Natriumhexameta- phosphat -	0,036 (0,001 n) 0,32,25 - - 1,0 -
Zirkondioxid	5,5-5,7	Wasser Wasser Wasser Wasser	Ölskure Tetranatriumpyrophosphat Kalium/Natriumhexameta- phosphat Natriumhexametaphosphat	o.A. 0,31,5 1,0 o.A.
Zirkonsilika't ·	4.7	Wasser	Kalium/Natriumhexameta- phosphat + Natriumcarbonat	0,6 0,12
Zucker (s.a. Robrzucker)	1,6	Isobutanol Diethylphthalat Isoamylalkohol Cyclòhexanon Isopropanol		-

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